

Structural Characterization and Low Temperature Transport Properties of FeSb₂ Single Crystals Grown by Vapor Transport

R. Vasquez^{1, 2, 3}, A. Kiswandhi^{1,4}, L. Winter^{1,4}, J. Whalen¹, A. Ramirez⁵, J. Brooks^{1,4}, T. Siegrist^{1,2}

1) National High Magnetic Field Laboratory, 1800 Dirac Dr., Tallahassee, FL 32310

- ²⁾ Department of Chemical & Biomedical Engineering, Florida State University, 2525 Dirac Dr., Tallahassee, FL 32310
- 3) International Institute for Complex Adaptive Matter, University of California, 1 Shields Avenue, Davis, CA 95616
 - 4) Department of Physics, Florida State University, 77 Chieftain Way, Tallahassee, FL 32306
- 5) Baskin School of Engineering, University of California Santa Cruz, 1156 High St., MS SOE, Santa Cruz, CA 95064

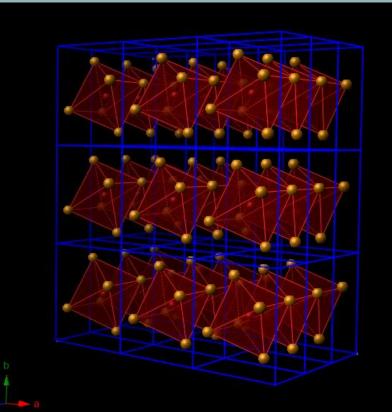
Abstract

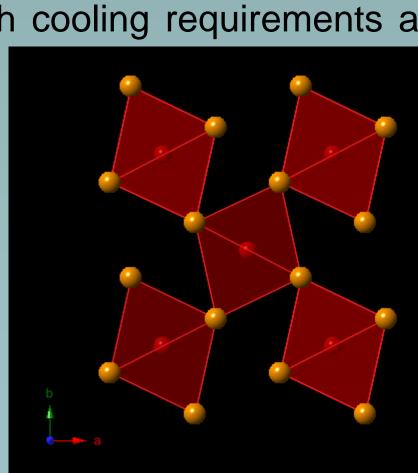
Bulk single crystals of iron di-antimonide, FeSb₂, are investigated for their transport properties at low temperatures. Crystals of sizes ranging 1.5-2.5 mm are grown through chemical vapor transport (CVT) using Cl₂ as transporting agent. Cl₂ is supplied to the reaction via decomposition of FeCl₃ at high temperatures. p(T) shows semiconducting behavior with the formation of a plateau at 10K-30K, along with a hybridization gap (ε_α~6.9-7.2meV) consistent with intrinsic behavior observed in correlated materials. Resistivity increases from room temperature (RT) by 5-6 orders of magnitude, higher than previous reports. Washing crystals achieved a reduction of Cl₂ concentration and did not appear to suppress the resistivity plateau, but did result in lower overall $\rho(T)$ increase from RT and a change in ϵ_{α} values.

Introduction

FeSb₂ has emerged as a potential low temperature thermoelectric material with Peltier cooling applications. It has been reported that FeSb₂ shows significantly enhanced thermoelectric properties at temperatures below 77K[1]. This underscores its potential for helium-free cooling for superconductor applications and others with cooling requirements as low as 10-15K.

FeSb₂ is a narrow band gap semiconductor electron with strong correlation behavior, similar to other iron-based semiconductors[2]. The narrow band gap in FeSb₂ is believed to originate from hybridization of Fe 3d orbitals with Sb 5p and 5s states, due to lattice distortions caused by edgesharing FeSb₆ octahedra formed along the long axis of the unit cell[3]. The





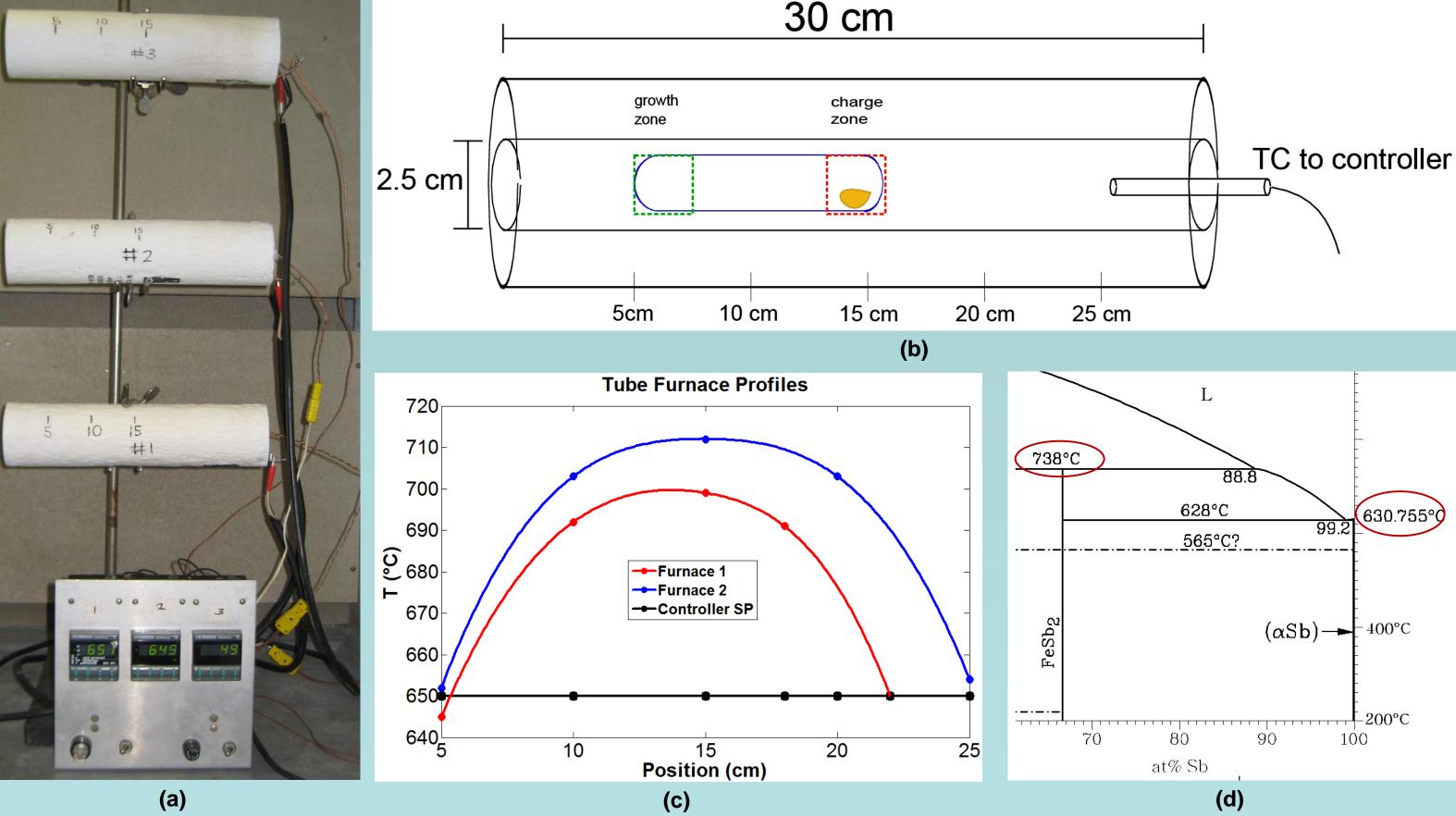
(a) Edge-sharing Fesb₆ octahedral chains along c-axis. (b) Tilted cornersharing FeSb₆ octahedra along a-b plane

enhancement of thermoelectric properties has been associated to the activation of the hybridization gap (~6meV[1]) in the material, accompanied by large carrier mobility.

CVT Growth of FeSb₂ Single Crystals.

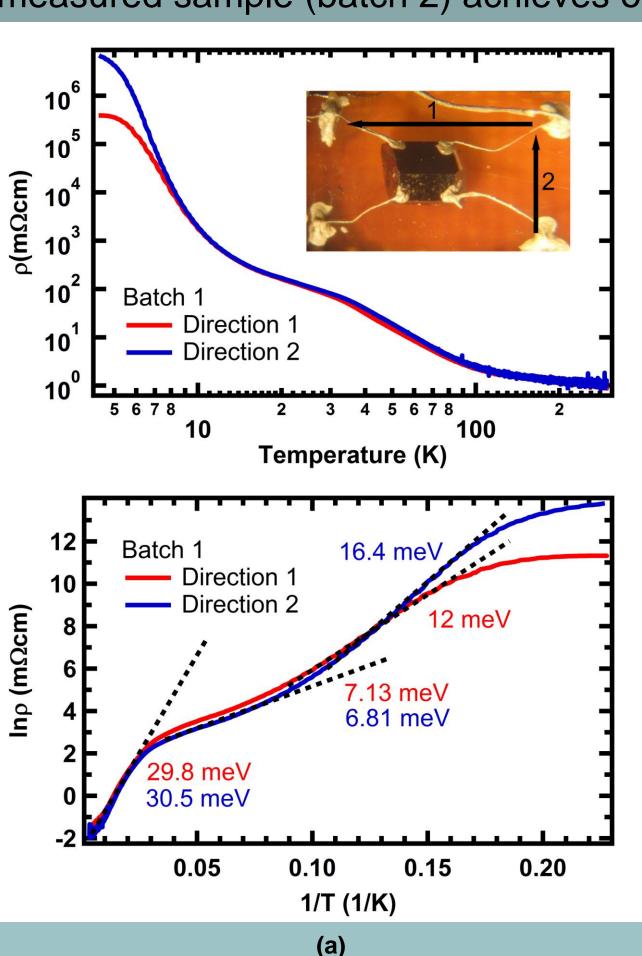
CVT Reaction Parameters Back-transport Hot (Charge) Zone 700 °C 650 °C Cold (Growth) Zone **Reaction Time** 5 days Cl_{2(a)} Pressure

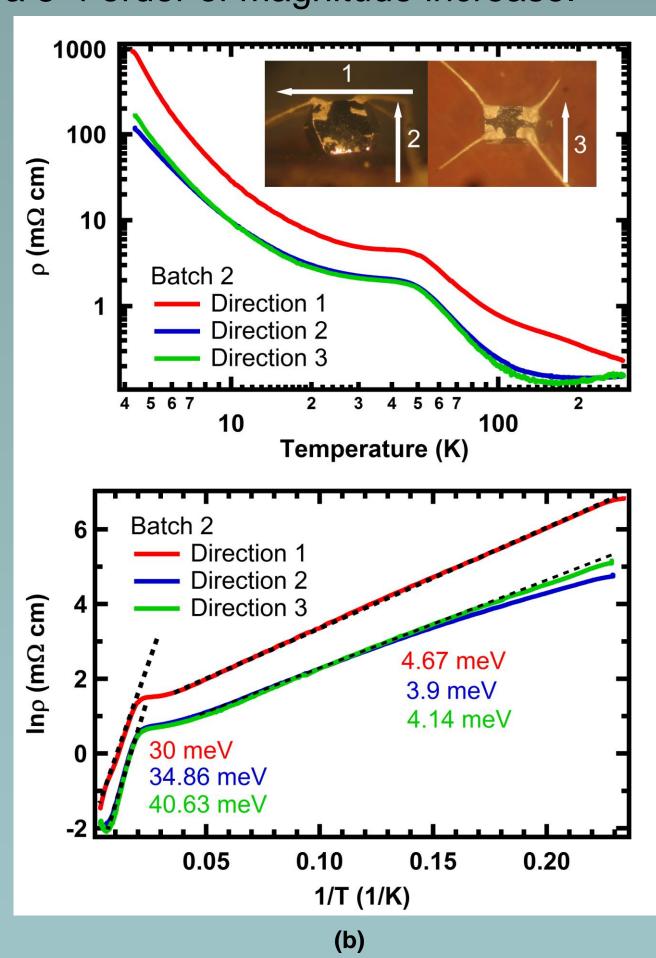
FeSb₂ growth carried out with stoichiometric amounts of elemental Sb (Sigma Aldrich, 99.9998%) and Fe (Alfa Aesar, 99.99%); mixed with anhydrous FeCl₃ (Sigma Aldrich, 97%). FeCl₃ decomposition above 315°C provided $Cl_{2(n)}$ for Fe transport into the vapor phase [4]. Crystals of sizes ranging 1.5-2.5mm were synthesized.



Transport Properties in FeSb₂

 $\rho(T)$ vs. T shows semiconducting behavior in FeSb₂, with weak temperature dependence in the range 10K-30K, as observed previously[1,2]. The overall magnitude of $\rho(T)$ increase with respect to room temperature (RT) shows sample dependence. For the first sample (batch 1) the observed increase relative to RT is nearly 2 orders of magnitude higher than previously reported[1,2,3]. A second measured sample (batch 2) achieves only a 3-4 order of magnitude increase.

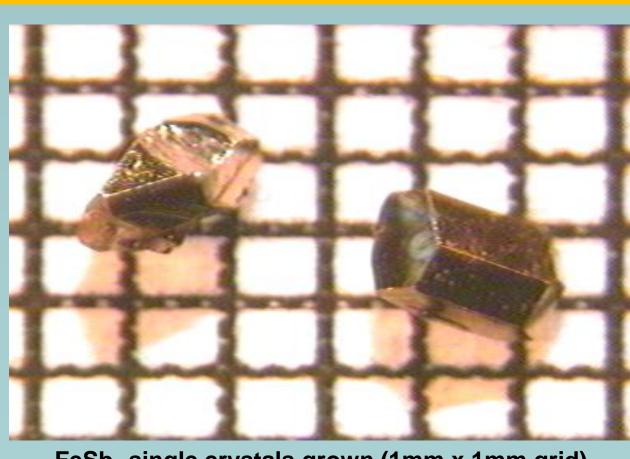




(a) Unwashed FeSb₂ crystal shows 5-6 order of magnitude increase in $\rho(T)$, with ϵ_g consistent with previous reports[1,2]. (b) FeSb₂ crystal washed in water and organic solvents to remove FeCl₃ excess shows decrease in ρ(T) at low temperatures and a change in ε_{σ} values.

Structural and Chemical Characterization of Crystals

Single crystal x-ray diffraction experiments an Oxford Instruments conducted with Xcalibur™2 4-Circle Kappa Single Crystal Diffractometer, Mo K α 1 (λ =0.71073 nm) source. SEM-EDS measurements conducted in a JEOL® JSM-5900 scanning electron microscope. Powder x-ray diffraction analysis confirmed phase purity of the crystals, with no secondary crystalline phases present. EDS analysis confirms FeSb₂ stoichiometry, with less than 0.20 at% CI impurity.



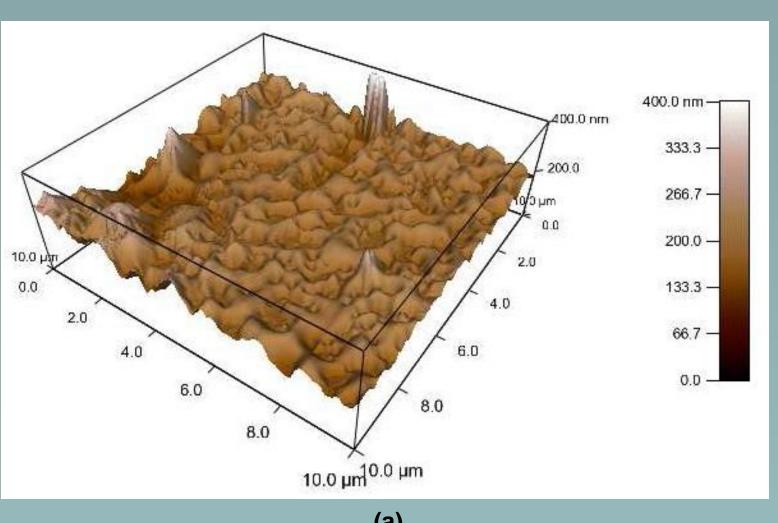
FeSb₂ single crystals grown (1mm x 1mm grid)

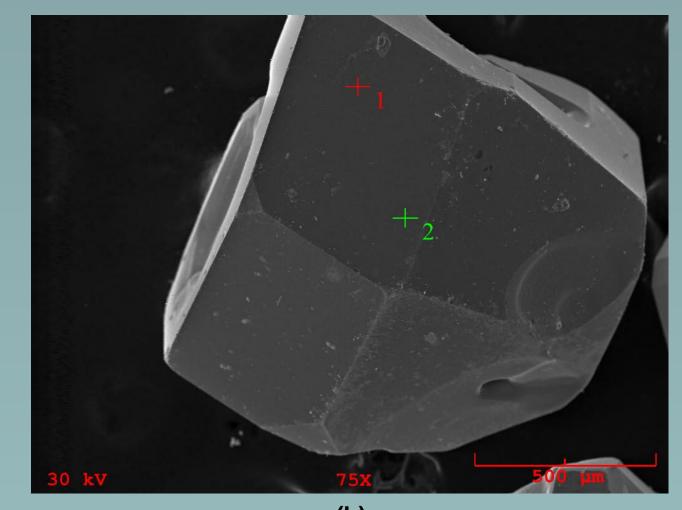
Single Crystal Diffraction Results Pnnm (N. 68) Space group Reference[6] This work a = 5.8306(1)a = 5.829(1)**Lattice parameters (Å)** b = 6.5380(1)b = 6.535(1)c = 3.1977(1)c = 3.196(1) $V(Å^3)$ 121.90 121.74 8.156 Density (g/cm³) 27.144 μ (mm⁻¹) Reflections collected 7073 0.0450 $3.1095 < \theta < 66.1663$ Data collection range (°) **Atomic Positions** 0.0000 0.0000 0.0000 0.0000

Topography in CVT-Grown FeSb₂ Bulk Single Crystals

Topographic studies conducted with an Asylum Research MFP-3D Atomic Force Microscope. The growth regime deviates from ideal layer-by-layer growth and instead appears to follow surface roughening behavior with multiple surface nuclei[7].

FeSb ₂ Surface Roughness		
Surface Characteristics	Surface Area (µm²)	RMS (nm)
Scattered elevations	25	17.896
Flat valley	0.09	0.442





(a) FeSb₂ crystal surface featuring valleys and elevations. Larger peaks (~400 nm) possibly related to surface impurities or small cracks and irregularities. (b) SEM image of FeSb₂ single crystal showing possible irregularities on the surface.

Conclusions and Future Work

- FeSb₂ single crystals were successfully grown by chemical vapor transport using Cl₂ (via FeCl₃ decomposition above 315 °C) as transporting agent. Structural and chemical characterization via x-ray diffraction (single crystal and powder) and SEM-EDS confirmed FeSb₂ stoichiometry without secondary crystalline phases present and Cl₂ impurity concentrations below 0.2 at%.
- ρ(T) measurements reveal semiconducting behavior in FeSb₂ with the formation of weak temperature dependent plateau in the range 10K-30K, consistent with what has been observed in previous studies. The magnitude of resistivity increase with respect to RT appears to be sensitive to post-CVT sample treatments.
- Topological studies on FeSb₂ single crystals show a growth environment deviating from smooth layered growth and following a surface roughening model during CVT. An RMS value of 0.44 nm was estimated for relatively flat areas on the surface.
- FeSb₂ samples will be further evaluated for low temperature thermoelectric performance, with Seebeck coefficient, thermal conductivity and charge carrier concentration studies currently underway. Future work also focuses on the effect of Sb-site doping on the overall improvement of key thermoelectric parameters.

References & Acknowledgements

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